

DIHYDROCOMPACTIN,
A NEW POTENT INHIBITOR
OF 3-HYDROXY-3-METHYLGLUTARYL
COENZYME-A REDUCTASE
FROM *PENICILLIUM CITRINUM*

Sir:

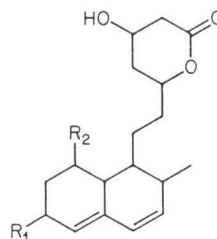
Compactin (Fig. 1) was first isolated from *Penicillium brevicompactum* by BROWN *et al.*¹⁾ as an antifungal metabolite. Later, ENDO *et al.*²⁾ reported their isolation of compactin (designated as ML-236B) together with ML-236A and ML-236C as hypocholesteremic agents from *Penicillium citrinum*. ENDO also reported monacolin K from *Monascus ruber*³⁾. Independently, the same compound (named mevinolin) was isolated by ALBERTS *et al.*⁴⁾ from *Aspergillus terreus*. 4a, 5-Dihydromevinolin (Fig. 2)* was also isolated from the same culture.⁵⁾ All of these five structurally-related entities are potent competitive inhibitors of 3-hydroxy-3-methylglutaryl (HMG)-CoA reductase (EC 1.1.1.34)²⁻⁷⁾, the rate-limiting enzyme in cholesterol synthesis, with mevinolin being the most potent. However, compactin is the most extensively studied. Its hypocholesteremic activity in several animal species including human subjects has been demonstrated⁸⁻¹²⁾, and a review of its pharmacology was also published¹³⁾.

In our search for new microbial metabolites having HMG-CoA reductase inhibition and ultimately hypocholesteremic activities, we have regrown *P. citrinum* in our laboratories and discovered a new component, designated as 4a,5-dihydrocompactin. This new compound apparently cannot be derived from compactin stereospecifically without lengthy chemical conversions. The present communication describes the isolation, physical and chemical properties of this unique compound. The inhibition of HMG-CoA reductase is also described.

P. citrinum (SANK 18767) was obtained as NRRL-8082¹⁴⁾ from the Northern Regional Research Laboratory, Northern Central Region, Agricultural Research Service, United States Department of Agriculture, Peoria, Illinois, U.S.A. It was grown aerobically in a medium containing 2% dextrose, 2% glycerol, 1% ardamine, 8 ppm CoCl₂·6H₂O, and 0.25% polyglycol

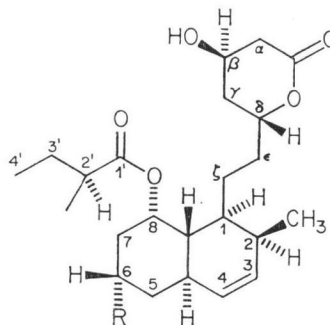
* The present numbering system conforms with the one used in ref. 4 for simplicity.

Fig. 1. Structures of the lactone forms of ML-236A, ML-236B (compactin), ML-236C and monacolin K (mevinolin).



ML-236A	R ₁ = -H,	R ₂ = -OH
ML-236B	R ₁ = -H,	R ₂ = -OCOCH(CH ₃)CH ₂ CH ₃
ML-236C	R ₁ = -H,	R ₂ = -H
Monacolin K	R ₁ = -CH ₃ ,	R ₂ = -OCOCH(CH ₃)CH ₂ CH ₃

Fig. 2. Structures of the lactone forms of 4a,5-dihydrocompactin and 4a,5-dihydromevinolin.



4a,5-Dihydrocompactin R = -H

4a,5-Dihydromevinolin R = -CH₃

P2000 at pH 7.0 for 4 days. The culture filtrate (10 liters) was extracted with ethyl acetate (5 liters) at pH 4 (adjusted with 1 M HCl). The ethyl acetate layer was extracted with 0.5 M NaOH (2 × 2 liters). After being adjusted to pH 4 with 1 M HCl, the combined NaOH solutions were extracted with ethyl acetate (2 liters and 1 liter). The combined ethyl acetate layers were dried over Na₂SO₄, filtered, and then evaporated to dryness *in vacuo*. A toluene solution of the oily residue was refluxed for one hour to facilitate lactonization. After filtration, the toluene solution was evaporated to dryness *in vacuo* and the residue was dissolved in 18 ml hexane - toluene - methanol (4: 1: 1) and purified by partition chromatography in a Sephadex LH-20 column (3 × 40 cm) equilibrated in the same solvent system. After eluting with 300 ml of solvent, a 10 ml

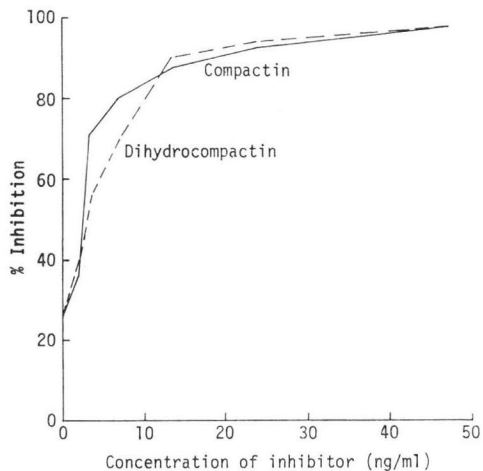
fraction was obtained which yielded an oil on concentration to dryness *in vacuo*. Reverse phase high pressure liquid chromatography of this material on an ES Industries Chromegabond C-18 column (0.9 × 50 cm) using acetonitrile-water (60:40) as the mobile phase yielded 45 mg of homogeneous (by TLC and HPLC) 4a,5-dihydrocompactin.

Exact mass measurements of 4a,5-dihydrocompactin established its molecular formula to be $C_{23}H_{36}O_8$ (Calcd: 392.2560; Found: 392.2558). Except for the obvious difference in mass units, the mass spectrum of our compound compared well with that of monacolin K published in reference 3 and the data for compactin in reference 1. In addition to peaks at m/z 392 (M^+), 307 ($M-85$), 290 ($M-102$), and 272 ($M-120$), prominent peaks at 186 ($M-206$), 159 ($M-233$) and 145 ($M-247$) were observed. The FT-IR spectrum (KBr) displayed absorptions at 3424, 2965, 1724 (lactone C=O), 1704 (ester C=O), 1258 and 1239 cm^{-1} . The 1H -NMR spectrum ($CDCl_3$) showed resonances in complete agreement with the assigned structure. In comparison with those of compactin and dihydromevinolin, the following proton assignments were observed: δ 5.60 (1H, m, H-3), δ 5.43 (1H, bd, $J=10$ Hz, H-4), δ 5.20 (1H, dt, $J=2.5$ Hz, H-8), δ 4.73 (1H, m, H- δ), δ 4.39 (1H, m, H- β), δ 2.75 (1H, dd, $J=5, 18$ Hz, H- α_a), δ 2.61 (1H, ddd, $J=1.5, 4, 18$ Hz, H- α_b), δ 1.12 (3H, d, $J=7$ Hz, C-2'- CH_3), δ 0.90 (3H, t, $J=7$ Hz, H-4'), δ 0.85 (3H, d, $J=7$ Hz, C-2- CH_3). In particular, the transoid stereochemistry at the ring juncture of the decalin system was strongly favored by the appearance of H-8a as a doublet ($J=2.5$ Hz) of a triplet ($J=11$ Hz) at δ 1.22. A large coupling of 11 Hz between the diaxial protons H-4a and H-8a was also observed for dihydromevinolin⁹. The Rf values on TLC (E. Merck, Kieselgel 60) were 0.40 in ethyl acetate-dichloromethane (4:6) and 0.18 in hexane-acetone (1:1).

Rat liver HMG-CoA reductase preparation and assay were performed as described by ALBERTS *et al.*⁴⁾ Fig. 3 clearly shows the similarity in potency between the inhibition of HMG-CoA reductase by compactin and dihydrocompactin. The I_{50} for both Na^+ salts (open hydroxy-carboxylate) are approximately 2.5 ng/ml. In a separate experiment, the K_i 's of the ammonium salts were determined to be 3.7 nM for dihydrocompactin and 3.8 nM for compactin.

Fig. 3. % Inhibition of rat liver HMG-CoA reductase by Na^+ salts of compactin and dihydrocompactin.

The preparation and assay of HMG-CoA reductase were as described by ALBERTS *et al.*⁴⁾



The compactin used as reference in this work was isolated from the same culture as dihydrocompactin in our laboratories and fully characterized to be identical to that described in references 1 and 14 by IR, NMR, UV and mass spectrometries, optical rotation, elemental analysis, and m.p. (149~150°C).

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